# Course Information Chemistry 1311-001 Summer I 2013

## **Contact Information**

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#### **Course Information**

Department: Physical Sciences

Start Date: 06/10/2013 End Date: 07/11/2013 Modality: Face-to-Face

Credits: 3 hour

Lecture: MTWTH 09:20 - 11:35 AM

### Required Reading/Material

Chemistry: The Central Science, 12<sup>th</sup> edition, Brown, T, LeMay, H, Bursten, B, Murphy, C, Woodward, P. Prentice Hall, 2012.

Simple Scientific calculator

(Programmable calculators cannot be used on quizzes or exams)

#### **Course Description**

The overall goal of the General Chemistry courses is to teach students to think about the properties and behavior of the macroscopic world in terms of the structure and arrangement of the constituent molecules and atoms. Specifically in CHEM 1311, aspects of atoms and quantum mechanics are emphasized through problem-solving questions.

#### <u>ICOs</u>

1, 2, 3

#### **Learning Outcomes**

Upon successful completion of this course, students will:

- 1. Define the fundamental properties of matter.
- 2. Classify matter, compounds, and chemical reactions.
- 3. Determine the basic nuclear and electronic structure of atoms.
- 4. Identify trends in chemical and physical properties of the elements using the Periodic

  Table.
- 5. Describe the bonding in and the shape of simple molecules and ions.
- 6. Solve stoichiometric problems.
- 7. Write chemical formulas.
- 8. Write and balance equations.
- 9. Use the rules of nomenclature to name chemical compounds.
- 10 Define the types and characteristics of chemical reactions.
- 11. Use the gas laws and basics of the Kinetic Molecular Theory to solve gas problems.
- 12. Determine the role of energy in physical changes and chemical reactions.
- 13. Convert units of measure and demonstrate dimensional analysis skills.

#### **Course Attendance**

Attendance is required and will be checked. To effectively master the material, your **attendance** for each class is necessary.

## **Academic Honesty**

Students are responsible for understanding the code of Student Conduct found in the OC Student Handbook. Academic Dishonesty will not be tolerated in this course. Cheating in any form will be handled accordingly.

#### **Students with Disability**

Odessa College complies with Section 504 of the Vocational Rehabilitation Act of 1973 and the ADA of 1990. Students with special needs or issues pertaining to access and participation in this class must contact me immediately. Further, you may call the Office of Disability services at 432-335-6861 to request assistance and accommodations.

### **Cell Phone**

When class begins, all phones need to be on silent or vibrate. If it is necessary for you to answer your phone or text message, then you need to step out into the hallway. If you consistently text message while in class, you will be asked to leave the class. Your phone **cannot** be used as a calculator on quizzes or exams.

#### **Course Grading**

Exams	60%
Quizzes	10%
Homework	10%
Final Exam	20%

#### **Homework**

Assignments will be made, collected and graded for completeness. The homework assignments are to help you prepare for the exams because variations of these will be on the exam. You should attempt to work all of the homework questions. Answer key will be posted the day before it's due. Homework will not be accepted late.

#### Quizzes

There will be a short quiz given once each chapter is completed. The quizzes will include questions and problems over the current material but may require some knowledge of the previous material. Your lowest quiz grade will be dropped at the end of the semester. If you miss a quiz, be prepared to make it up on the day you return to class, otherwise you will not be allowed to make it up.

#### **Examinations**

(tentatively)

First Exam: Monday, June 17<sup>th</sup>
Second Exam: Tuesday, June 25<sup>th</sup>
Third Exam: Tuesday, July2<sup>nd</sup>

**Final Exam:** Thursday, July 11<sup>th</sup>, class time

#### Make-Up Exam

A make-up exam will be given on the last day of class before finals. If you miss an exam for any reason you will have the opportunity to take this exam. Any student who may wish to replace a lower exam may also take this exam. The make-up exam will be multiple choice questions covering the entire semester.

#### Final Exam

The final exam is comprehensive.

### **HELP**

Please do not wait until the last minute to get help! Come in during office hours or make an appointment to see **ME** as soon as a concept or problem gives you difficulty. I am here to help you make it through this experience.

### **Expectations for Engagement – Face to Face Learning**

To help make the learning experience fulfilling and rewarding, the following Expectations for Engagement provide the parameters for reasonable engagement between students and instructors for the learning environment. Students and instructors are welcome to exceed these requirements.

Reasonable Expectations of Engagement for Instructors

- 1. As an instructor, I understand the importance of clear, timely communication with my students. In order to maintain sufficient communication, I will
  - provided my contact information at the beginning of the syllabus;
  - respond to all messages in a timely manner through telephone, email, or next classroom contact; and,
  - notify students of any extended times that I will be unavailable and provide them
    with alternative contact information (for me or for my supervisor) in case of
    emergencies during the time I'm unavailable.
- 2. As an instructor, I understand that my students will work to the best of their abilities to fulfill the course requirements. In order to help them in this area, I will
  - provide clear information about grading policies and assignment requirements in the course syllabus, and
  - communicate any changes to assignments and/or to the course calendar to students as quickly as possible.
- 3. As an instructor, I understand that I need to provide regular, timely feedback to students about their performance in the course. To keep students informed about their progress, I will
  - return classroom activities and homework within one week of the due date and
  - provide grades for major assignments within 2 weeks of the due date or at least 3 days before the next major assignment is due, whichever comes first.

Reasonable Expectations of Engagement for Students

1. As a student, I understand that I am responsible for keeping up with the course. To help with this. I will

- attend the course regularly and line up alternative transportation in case my primary means of transportation is unavailable;
- recognize that the college provides free wi-fi, computer labs, and library resources during regular campus hours to help me with completing my assignments; and,
- understand that my instructor does not have to accept my technical issues as a legitimate reason for late or missing work if my personal computer equipment or internet service is unreliable.
- 2. As a student, I understand that it is my responsibility to communicate quickly with the instructor any issue or emergency that will impact my involvement with or performance in the class. This includes, but is not limited to,
  - missing class when a major test is planned or a major assignment is due;
  - having trouble submitting assignments;
  - dealing with a traumatic personal event; and,
  - having my work or childcare schedule changed so that my classroom attendance is affected.
- 3. As a student, I understand that it is my responsibility to understand course material and requirements and to keep up with the course calendar. While my instructor is available for help and clarification, I will
  - seek out help from my instructor and/or from tutors;
  - ask questions if I don't understand; and,
  - attend class regularly to keep up with assignments and announcements.

## **Learning Resource Center (Library)**

The Library, known as the <u>Learning Resources Center</u>, provides research assistance via the <u>LRC's catalog (print books, videos, e-books)</u> and <u>databases (journal and magazine articles)</u>. <u>Research guides</u> covering specific subject areas, <u>tutorials</u>, and the <u>"Ask a Librarian"</u> service provide additional help.

#### **Student E-mail**

Please access your Odessa College Student E-mail, by following the link to either set up or update your account: http://www.odessa.edu/gmail/. All assignments or correspondence will be submitted using your Odessa College email.

#### **Student Portal**

Please access your Odessa College Student E-mail, by following the link to either set up or update your account: http://www.odessa.edu/gmail/. All assignments or correspondence will be submitted using your Odessa College email.

## **Technical Support**

For Blackboard username and password help and for help accessing your online course availability and student email account contact the Student Success Center at 432-335-6878 or online at https://www.odessa.edu/dept/ssc/helpdesk\_form.htm.

#### **Important School Policies**

For information regarding student support services, academic dishonesty, disciplinary actions, special accommodations, or student's and instructors' right to academic freedom can be found in the Odessa College Student Handbook.

#### **Design for Completion**

As part of the Design for Completion initiative, your Odessa College Student Success Coach and faculty mentor will help you stay focused and on track to complete your educational goals. If an instructor sees that you might need additional help or success coaching, he or she may submit a RetentionAlert. Your Student Success Coach or faculty mentor will contact you to work toward a solution.

## **Course Schedule**

(tentatively)

Chap	ter Topics	<u>Descriptions</u>	
I	Introduction: Matter and Measurement	How Science is Done	Quiz
II	Atoms, Molecules, and Ions	Atomic Theory	Quiz
III	Stoichiometry: Calculations with Chemical	What is a mole?	Quiz
	Formulas and Equations		
	One, Two, and Three		Exam
IV	Reactions in Aqueous Solutions	Types of Reactions	Quiz
V	Thermochemistry	Enthalpy	Quiz
VI	Electronic Structure of Atoms	Electron Arrangement	Quiz
	Four, Five, and Six		Exam
VII	Periodic Properties of the Elements	Trends of Periodic	Quiz
		Table	
VIII	Basic Concepts of Chemical Bonding	Bonding Types	Quiz
IX	Molecular Geometry and Bonding	Multiple Bonding	Quiz
	Theories		
	Seven, Eight, and Nine		Exam
X	Gases	Gas Laws	Quiz
	Final Exam		Exam

Good Luck this Semester!!!!!!!!

# CHEMISTRY 1311 Homework Set

## CHAPTER 1

1, 2, 4, 7, 8, 10, 12, 14, 17, 19, 20, 27, 31, 33, 35-43, 46, 48, 55, 58, 64, 71

## **CHAPTER II**

2, 4, 5, 6, 7, 10, 15, 19, 20, 22, 24, 26, 31, 32, 37, 38, 43, 45, 48, 50, 55, 56, 58, 61, 64, 66, 68, 70, 73, 74, 99

## CHAPTER III

1, 4, 5, 6, 7, 11, 12, 14, 16, 18, 22, 24, 34, 36, 38, 46, 50, 54, 62, 68, 77, 84, 93, 103

## \*Extra Credit Opportunity

## **CHAPTER IV**

2, 7, 12, 15, 16, 19, 20, 22, 24, 28, 29, 35, 37, 40, 47, 50, 52, 62, 63, 64, 71, 78, 80, 87, 99

# **CHAPTER V**

3, 8, 15, 27, 32, 39, 41, 43, 53, 55, 63, 66, 69, 71, 73, 85, 102

# \* Extra Credit Opportunity

## **CHAPTER VI**

2, 5, 13, 18, 23, 27, 35, 53, 65, 66, 68, 70, 71, 81

# **CHAPTER VII**

25, 26, 29, 30, 37, 41, 45, 47, 61, 68, 73, 81, 103 \*Extra Credit Opportunity

## **CHAPTER VIII**

4, 7, 13, 18, 19, 39, 40, 41, 42, 47, 48, 63, 69, 71, 85, 90

## **CHAPTER IX**

3, 7, 12, 14, 19, 27, 28, 32, 43, 44, 55, 56, 61, 66, 81, 83, 101

# **CHAPTER X**

4, 10, 21, 35, 37, 43, 53, 54